



# Iron Sulfur Molecular Inks for Pyrite Thin-Films

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Pyrite Collaboration Meeting

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# Outline

- Motivation
- Synthesis
- H<sub>2</sub>S annealing
- S<sub>2</sub> annealing
- Future Work

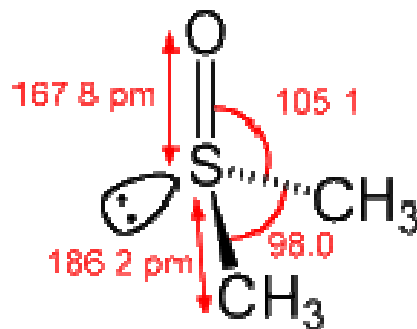


# Motivation for Molecular Inks

- **Goal:** Find a convenient method for film deposition that has the correct composition, phase purity, and grain structure
- Benefits of Using Molecular Ink approach:
  - Solution based approaches are easier to scale up
  - Well controlled stoichiometry and minimal to no contamination (elemental precursors)
  - Simple film deposition (spin casting, doctor blading, drop casting)

## Using dimethyl sulfoxide (DMSO) and ethanolamine as a solvent mixture to dissolve elemental precursors

- **Motivation:** Paper by Mitzi stating that a mixture of DMSO and ethanolamine was an effective wetting agent for thermally oxidized silicon substrates; less toxic than hydrazine
- **Previous molecular ink approaches:**
  - CZTS (copper zinc tin sulfur/selenide) films: Dissolved elemental precursors in hydrazine and obtained devices with over 9.6% efficiency (as opposed to previously obtained efficiencies of 3.2% and 6.7%)
  - CIGS (copper indium gallium sulfur/selenide): “ ” in hydrazine and obtained efficiencies of 10.1%



# Synthesis Process Inside N<sub>2</sub> Glovebox

- Each layer is roughly 150 nm
- 200 °C is ideal pre-baking temperature since the boiling points of DMSO and ethanolamine are 189 °C and 170 °C respectively; by FT-IR all organics seem to be removed

**Form mixture of DMSO  
and ethanolamine  
(6.5:1)**

**Dissolve S (1 M) in  
solution and stir for >6  
hours (at room temp)**

**Add Fe (0.33 M) and  
stir for >24 hrs.**

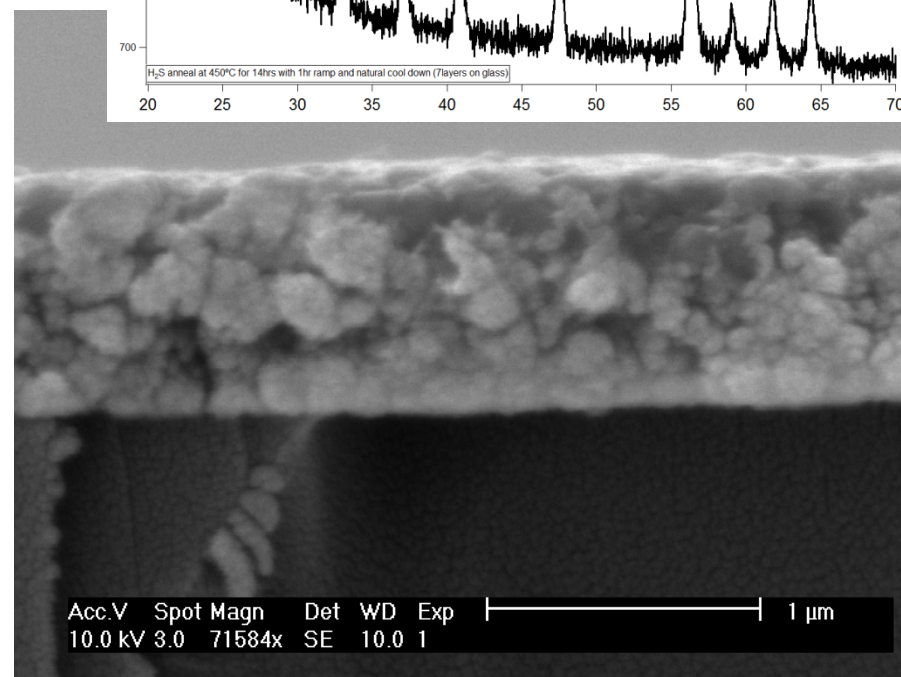
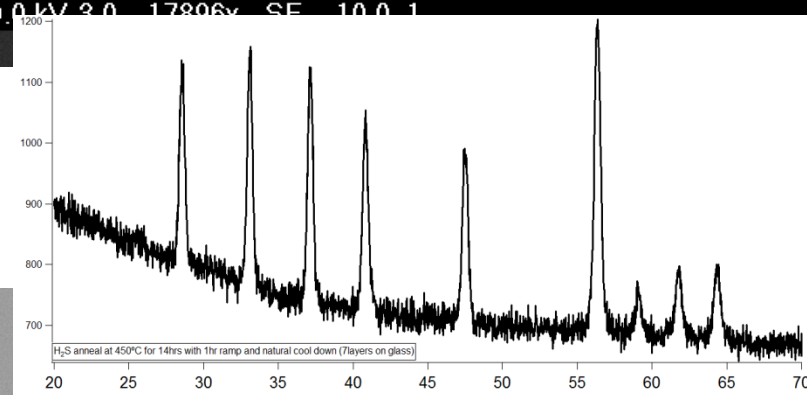
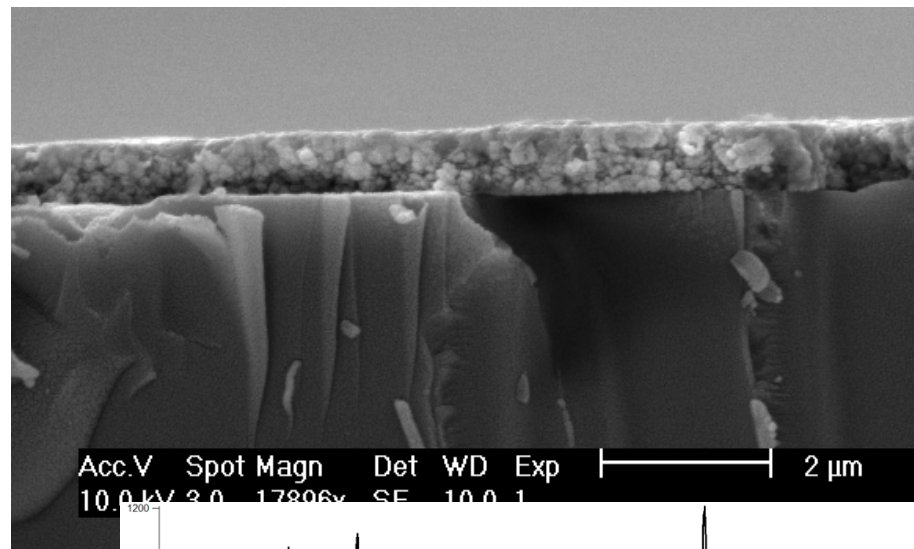
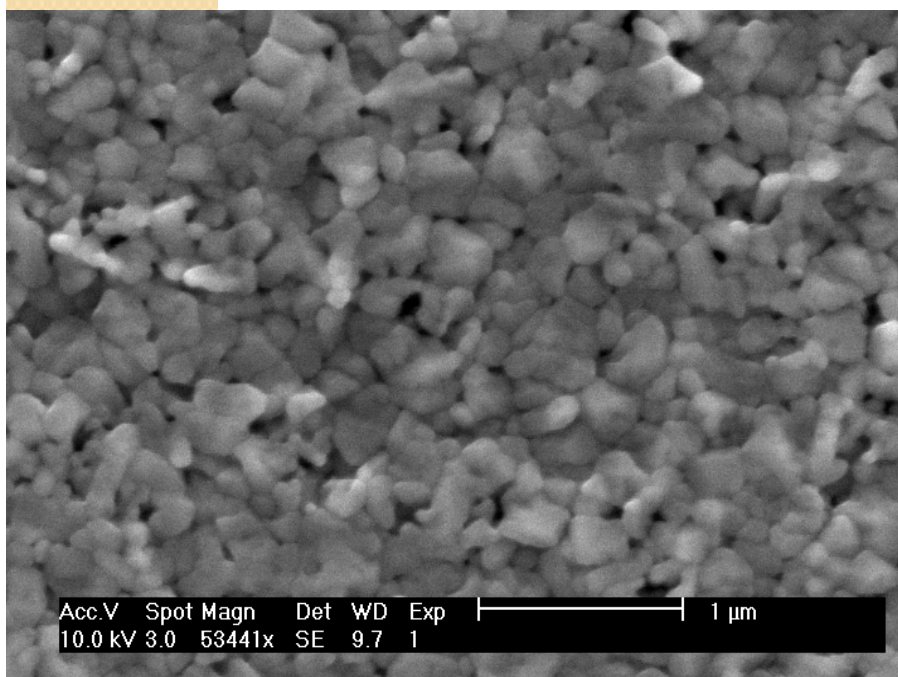
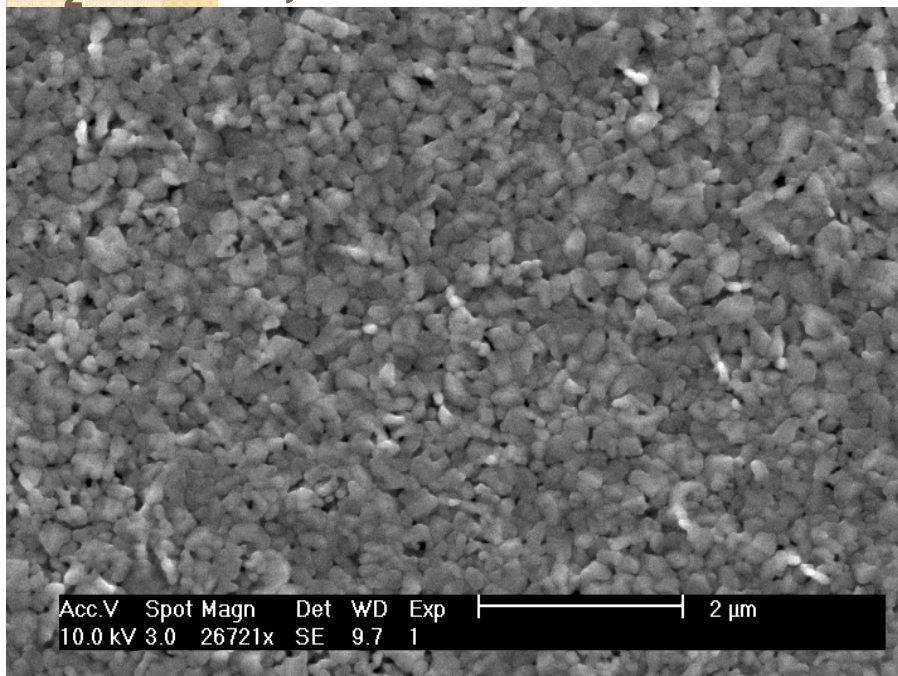
**Spin coat at 1500 rpm  
for 60s**

**Pre-bake on hotplate at  
200 °C for 5 min.**

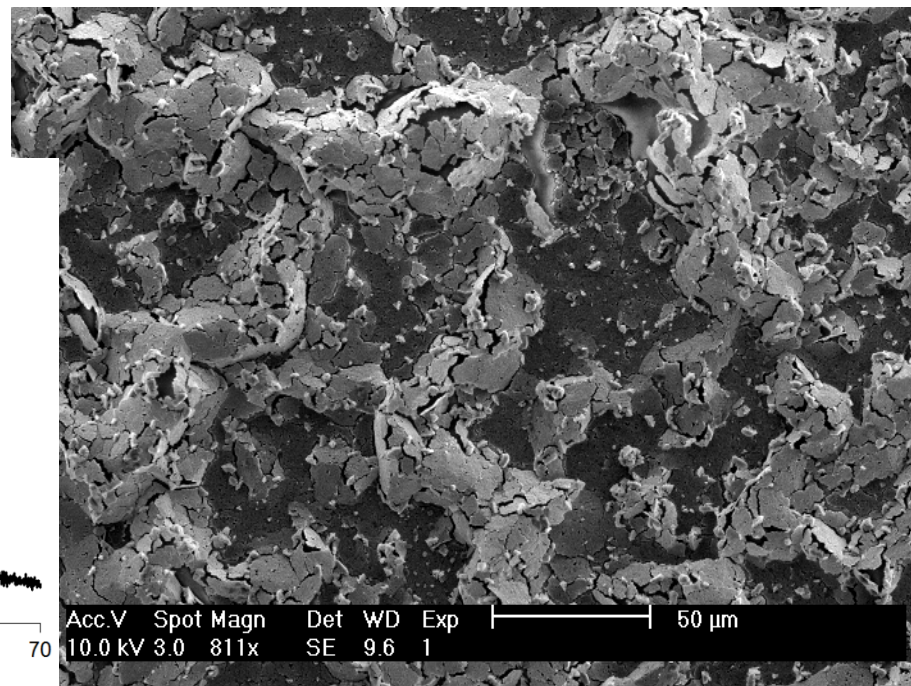
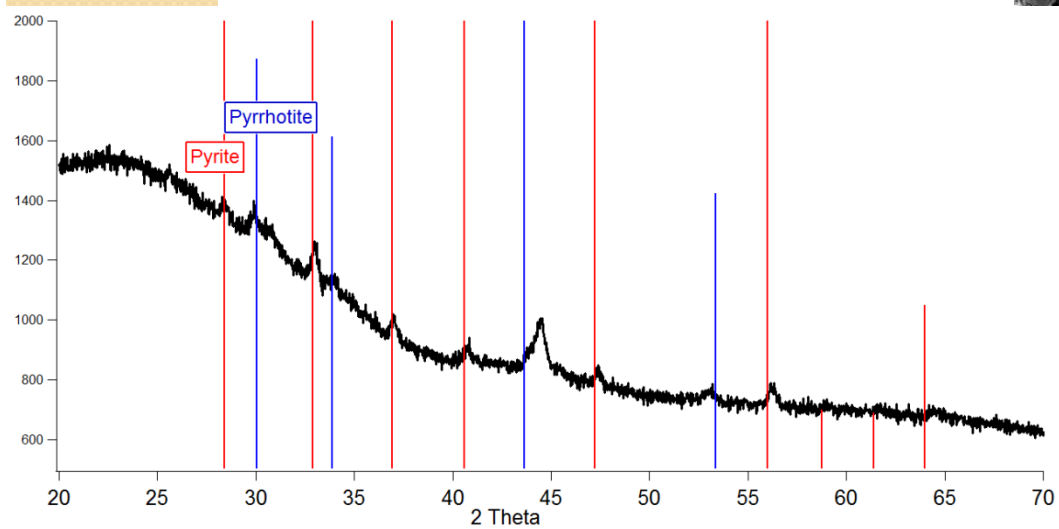
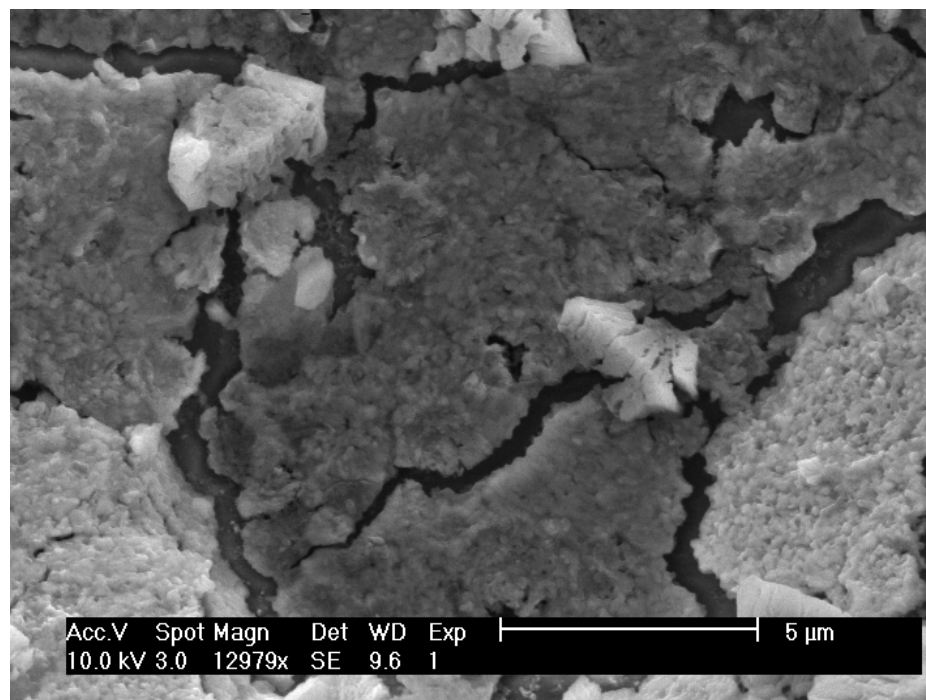
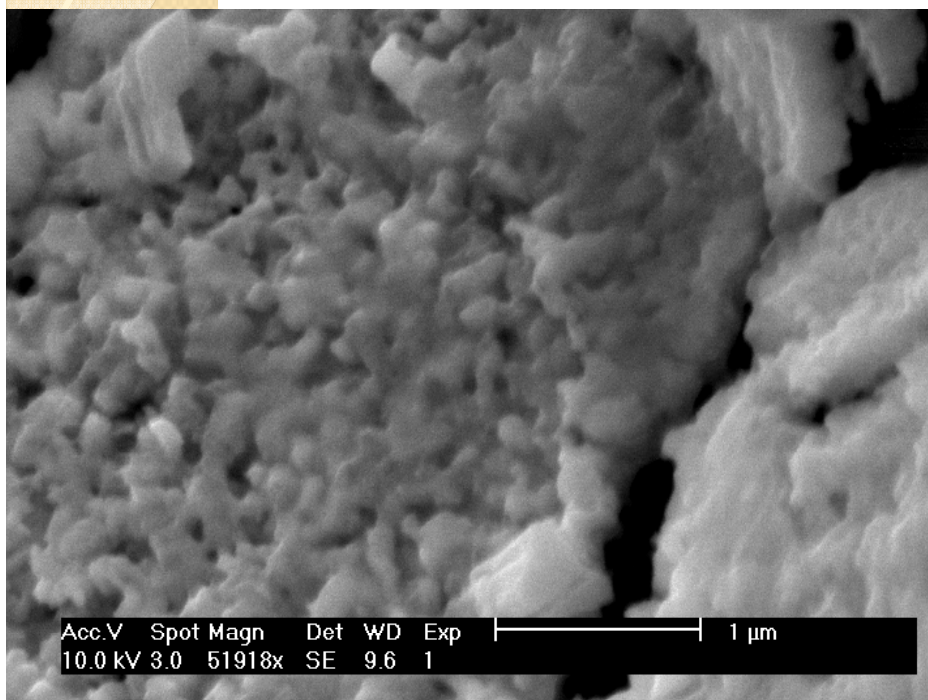
**Anneal in S (25 mg) at  
550 ° C for 6 hours**



# H<sub>2</sub>S 450 °C, 15hrs

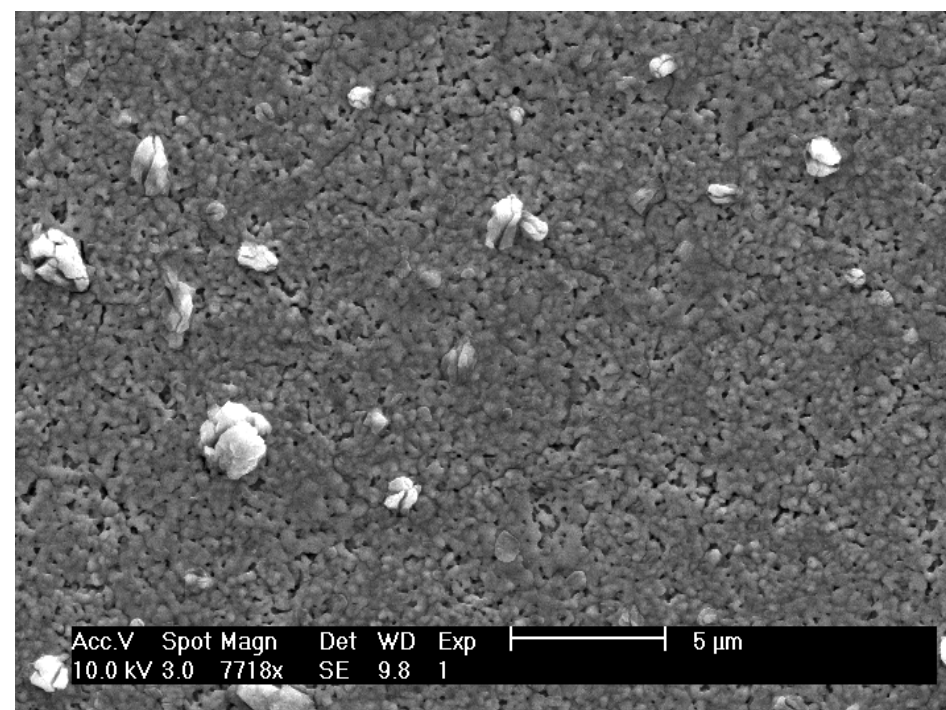
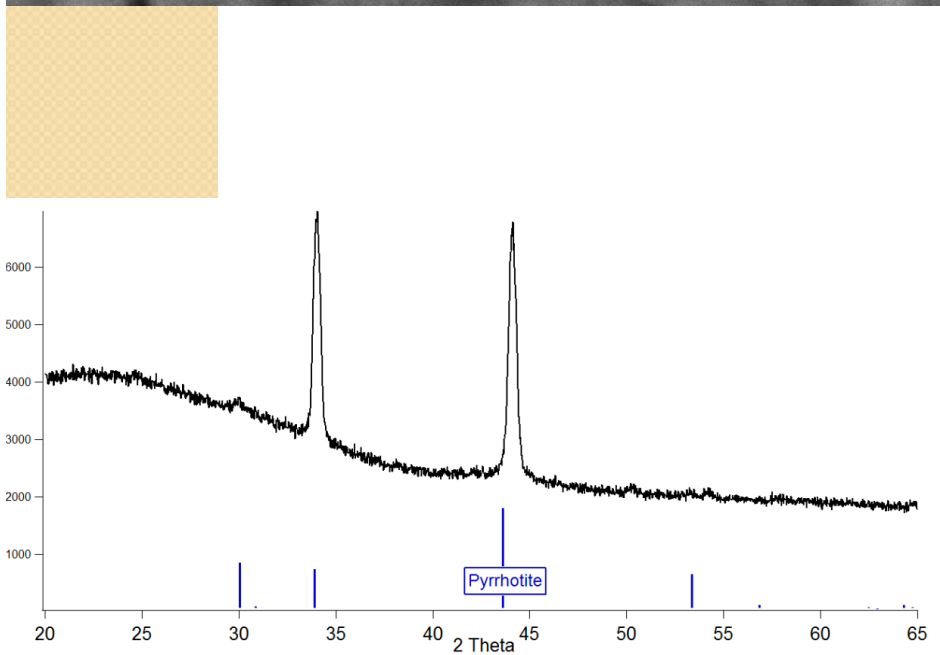
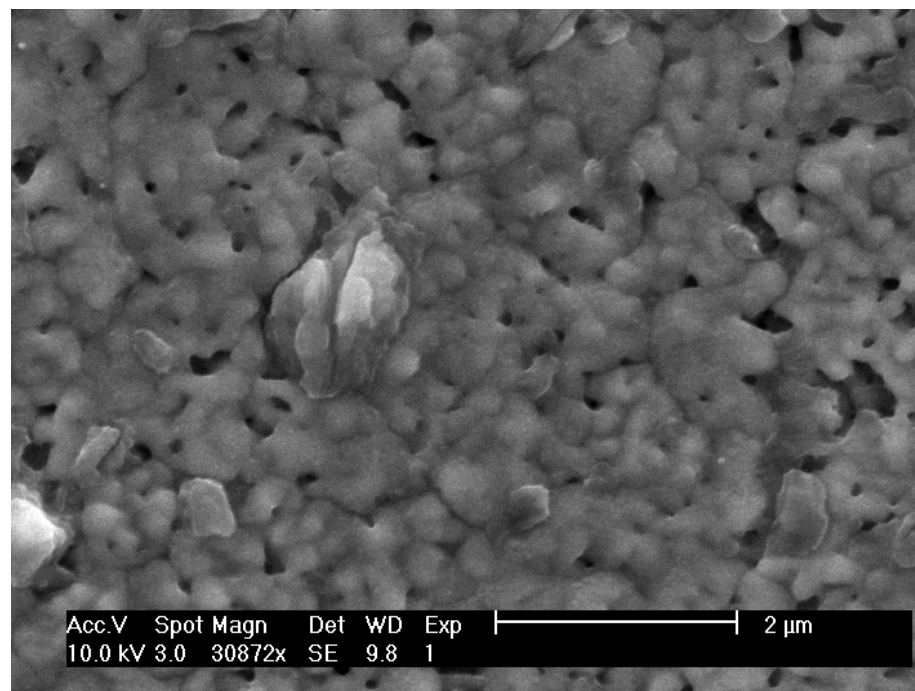
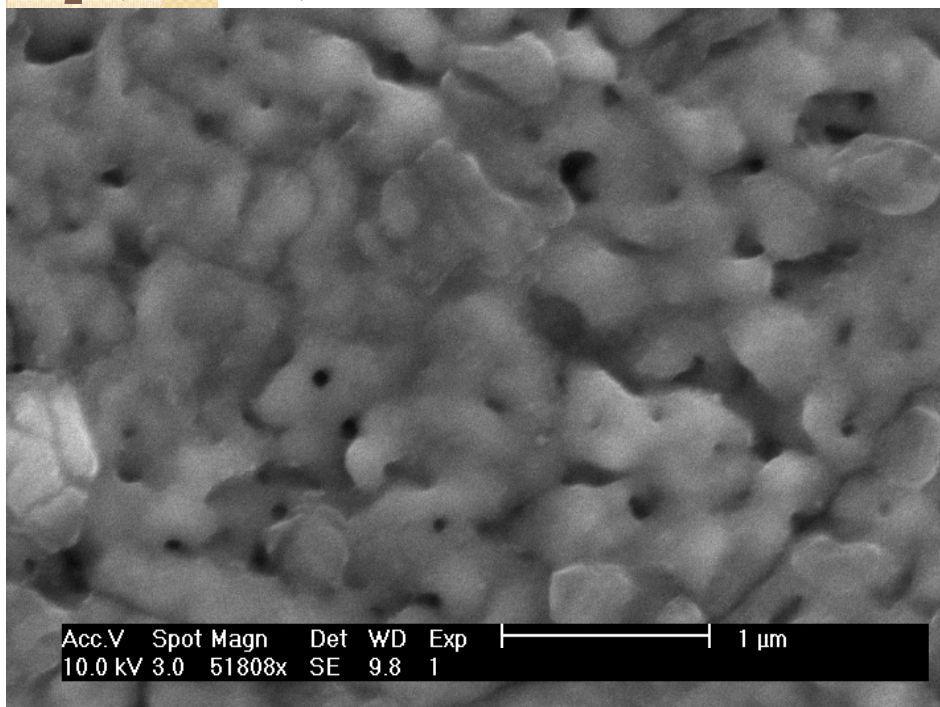


H<sub>2</sub>S, 500 °C, 15hrs



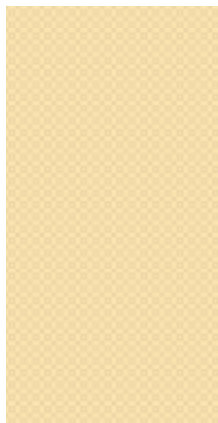
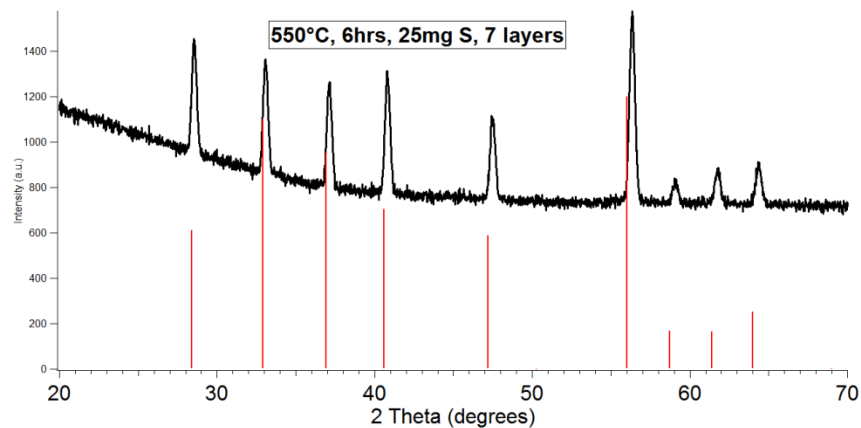
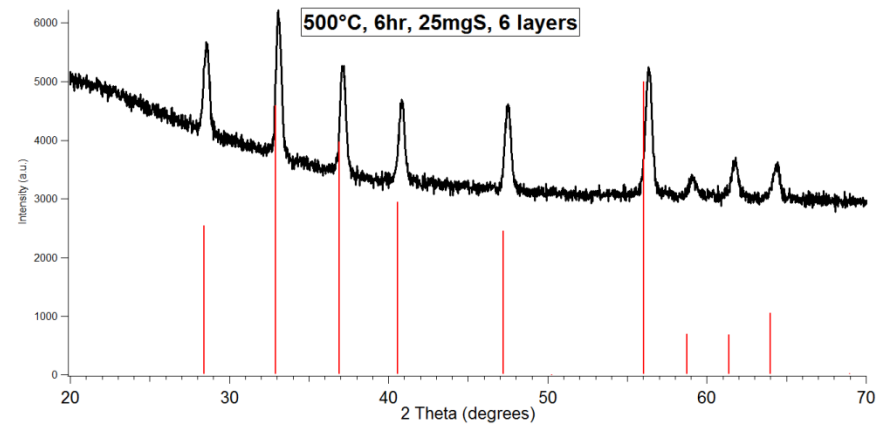
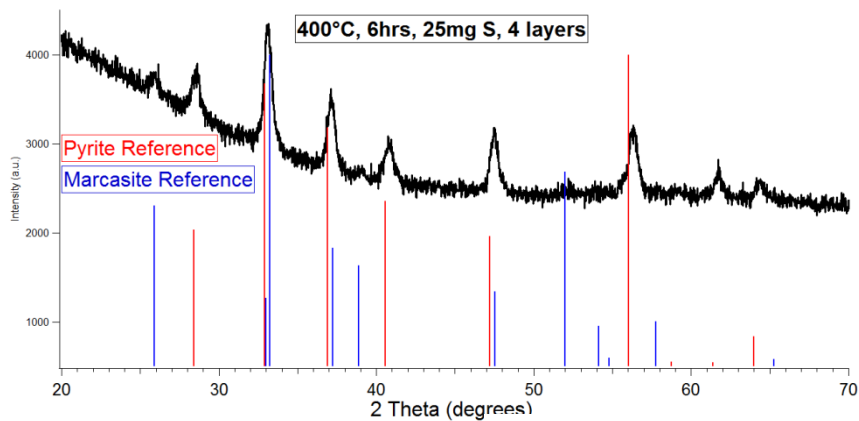
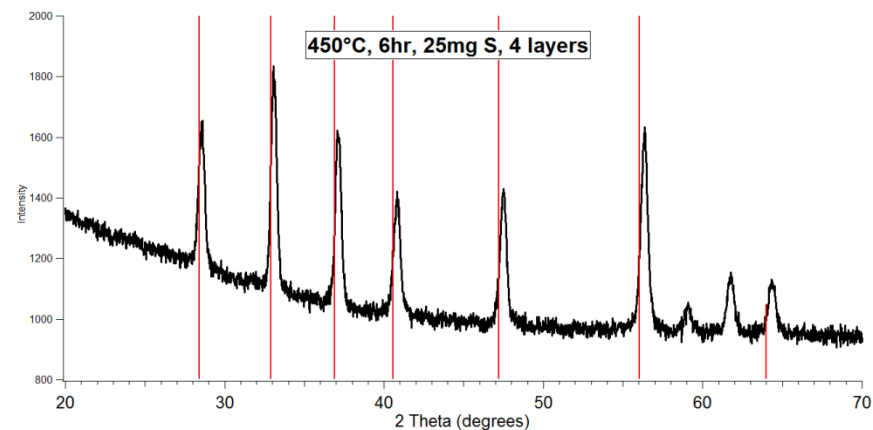
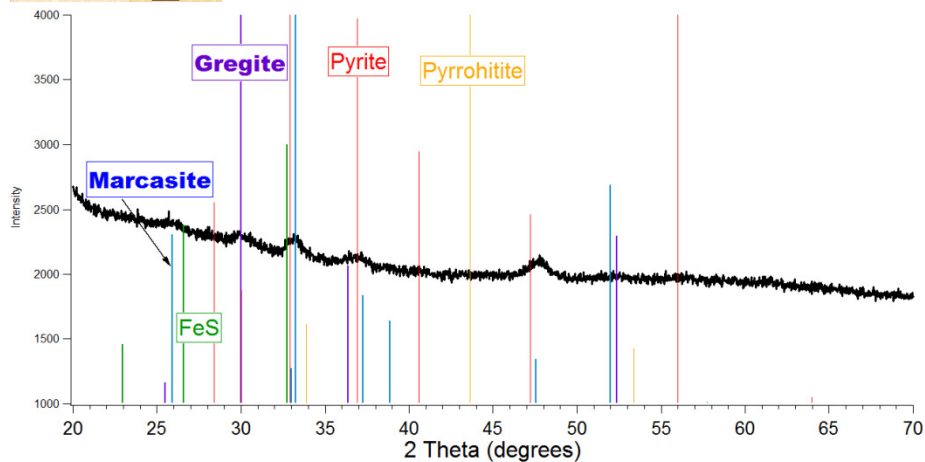


**H<sub>2</sub>S, 550 °C, 15hrs**



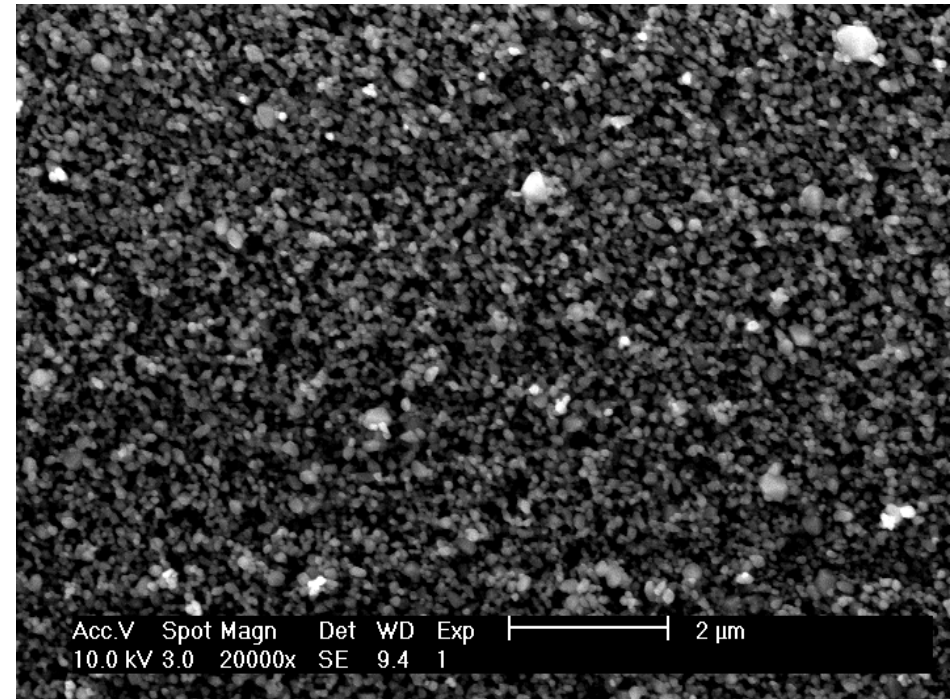
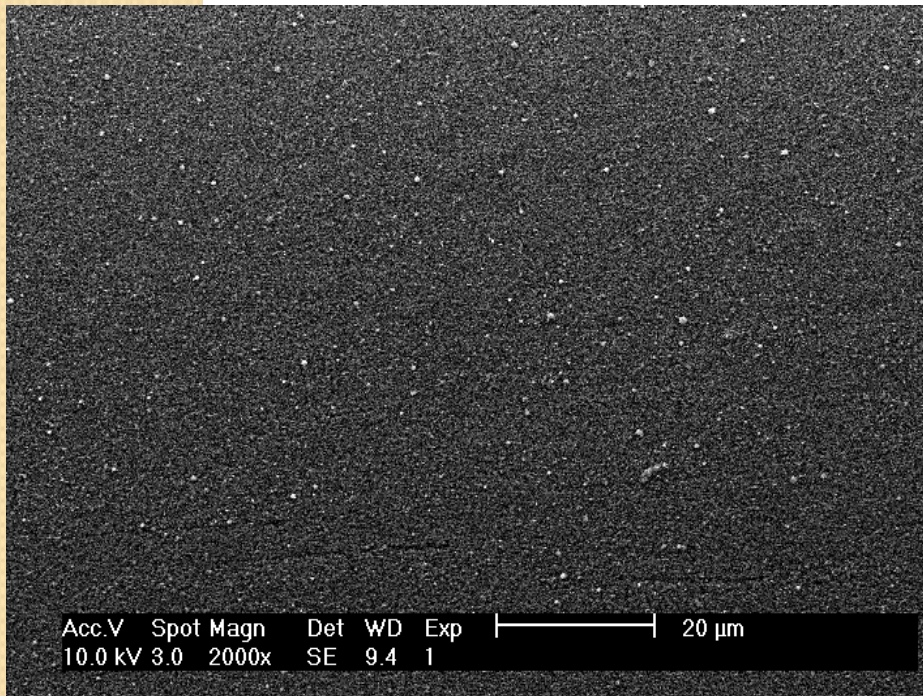
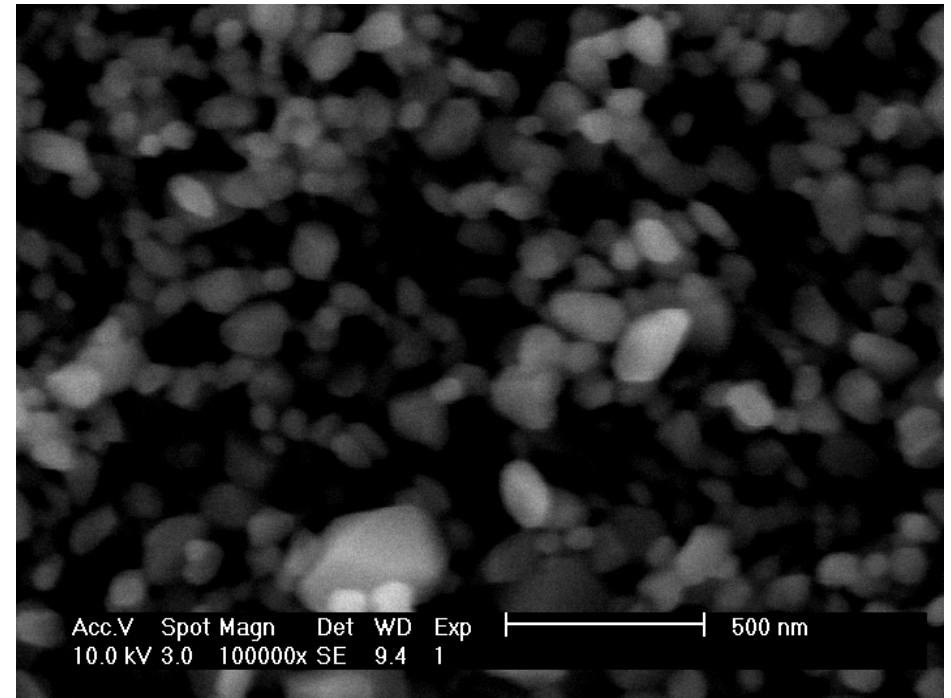


# S<sub>2</sub> Annealing X-Ray Diffraction Patterns



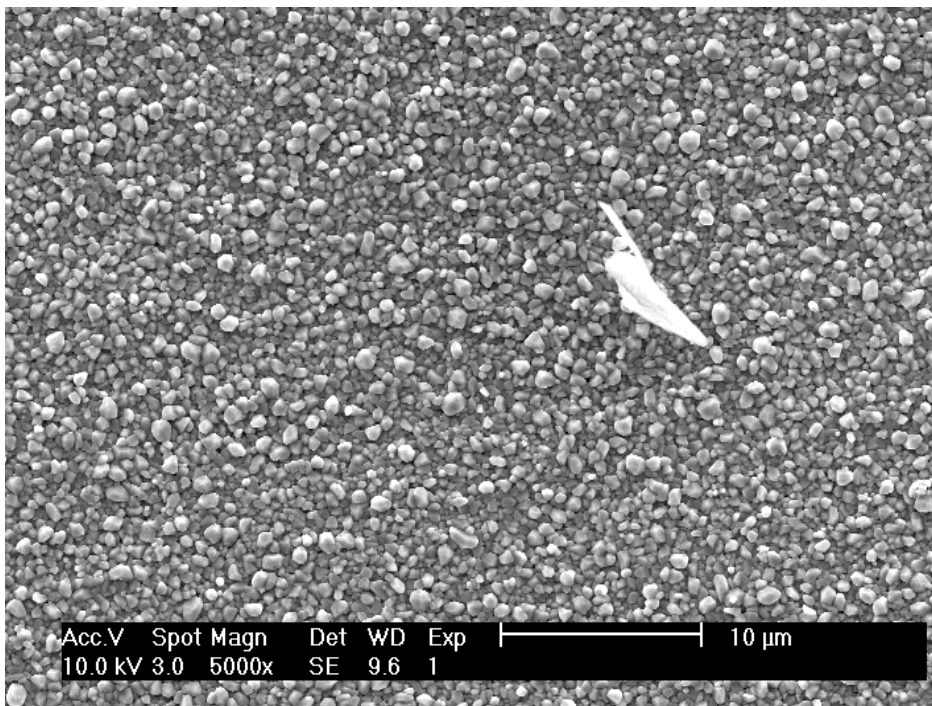
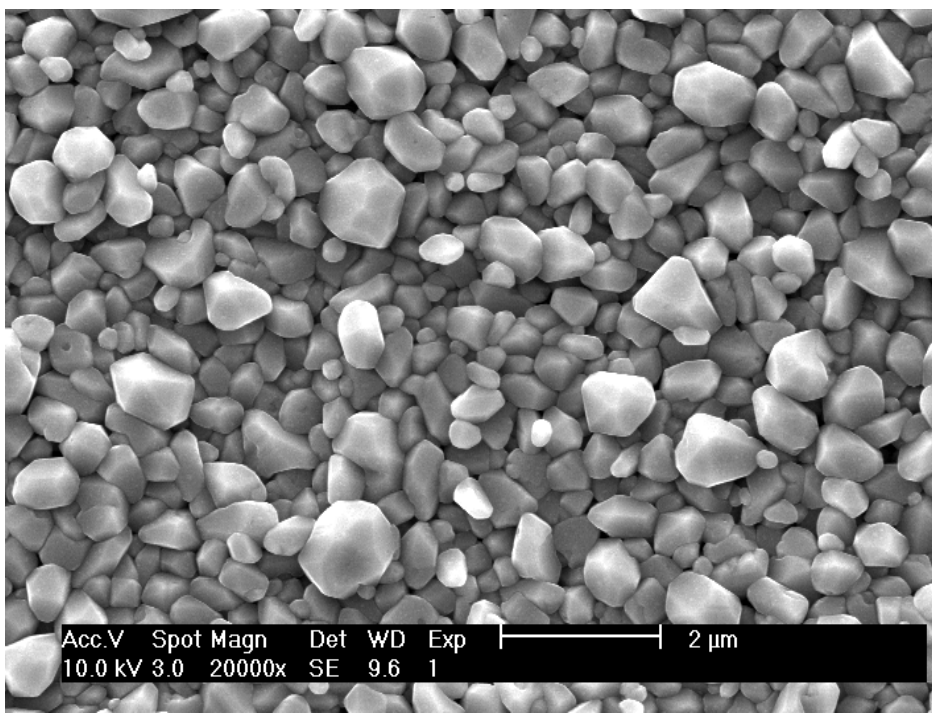
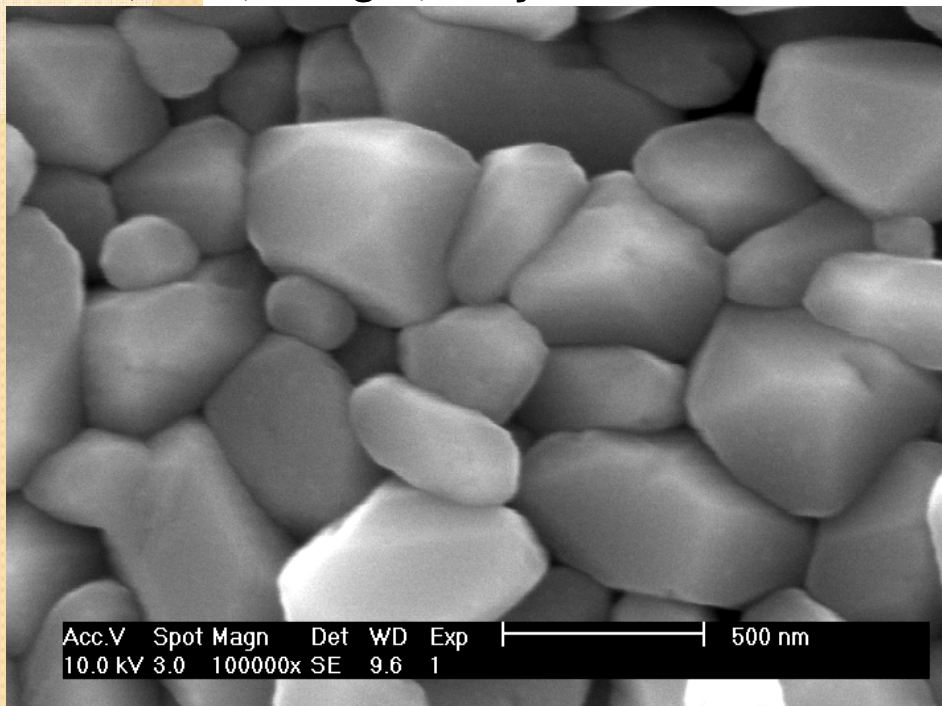
## 500 °C, 6hrs, 25 mg S

- Yields pure pyrite, but morphology is not ideal
  - Little grain growth
  - Not well connected grains



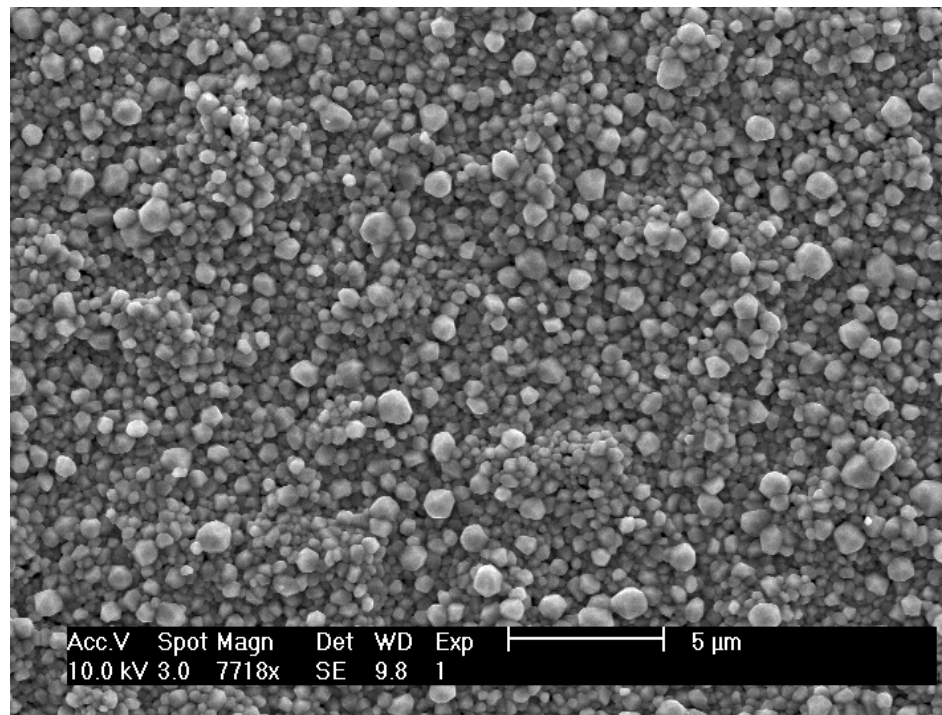
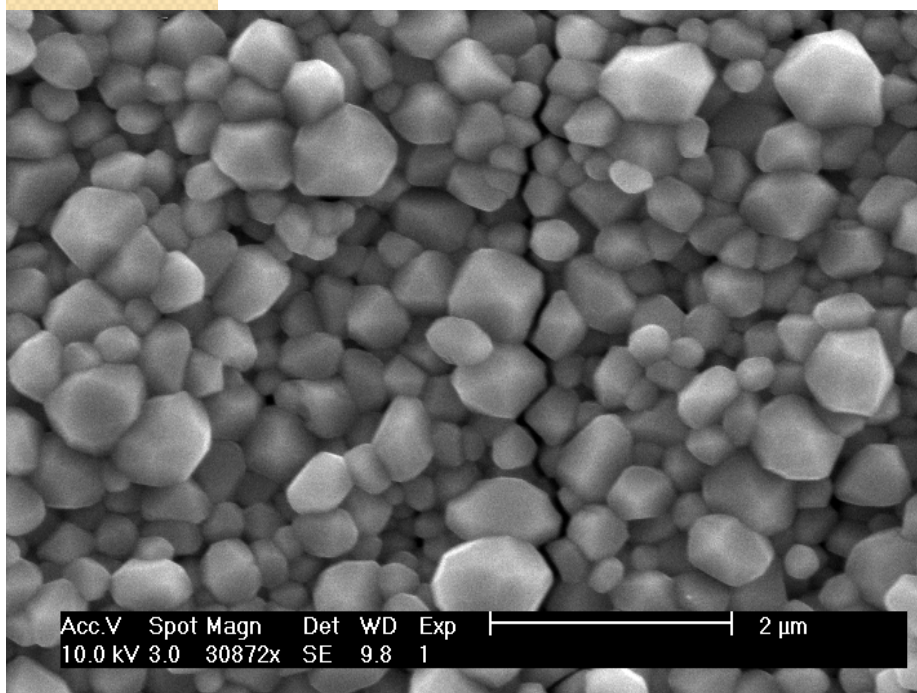
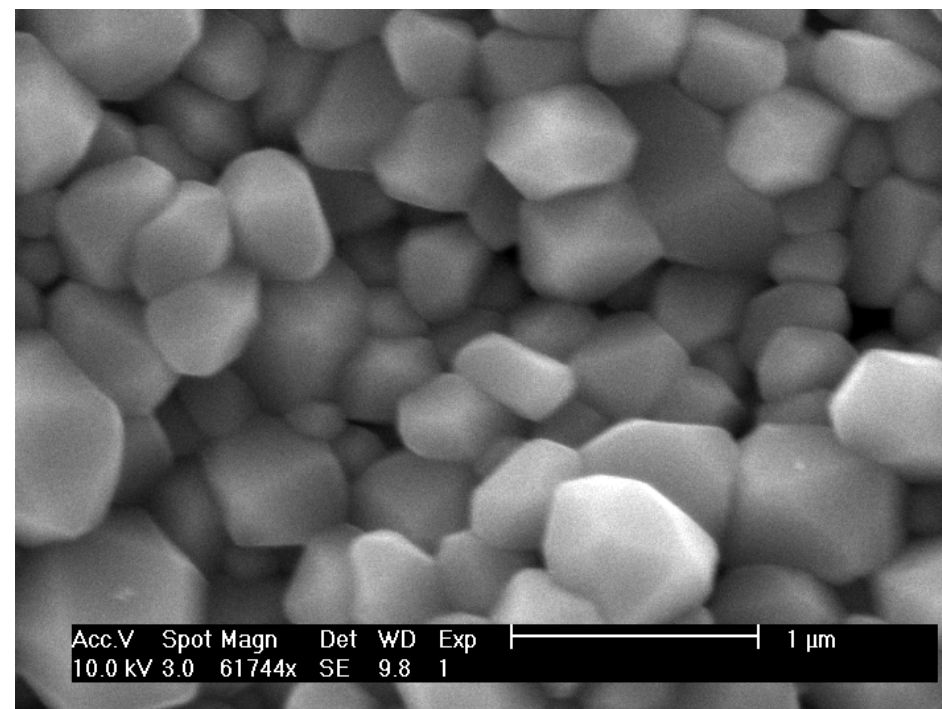
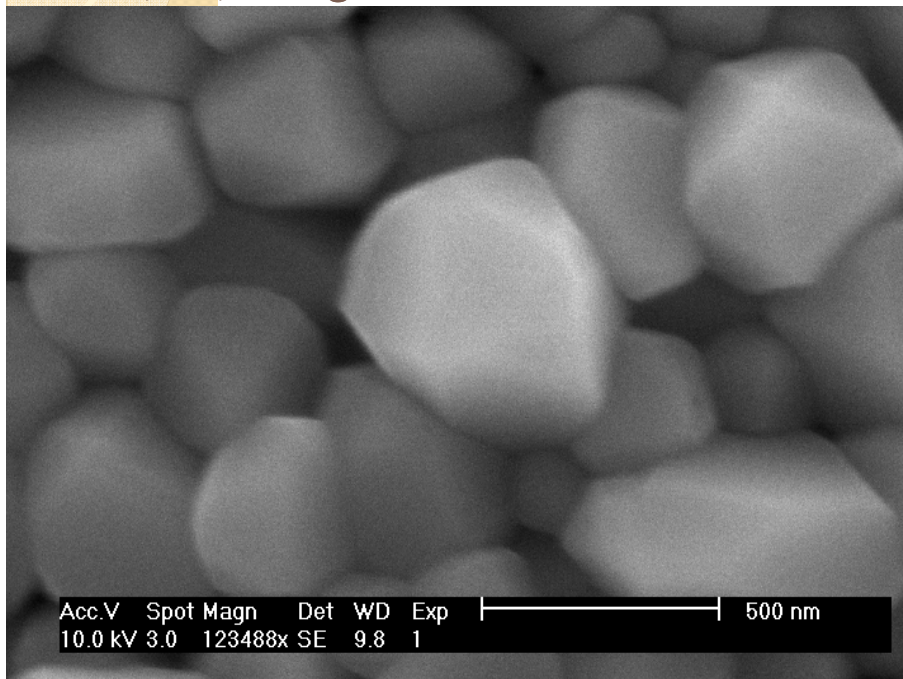


550 °C, 6hrs, 25mg S, 7 layers



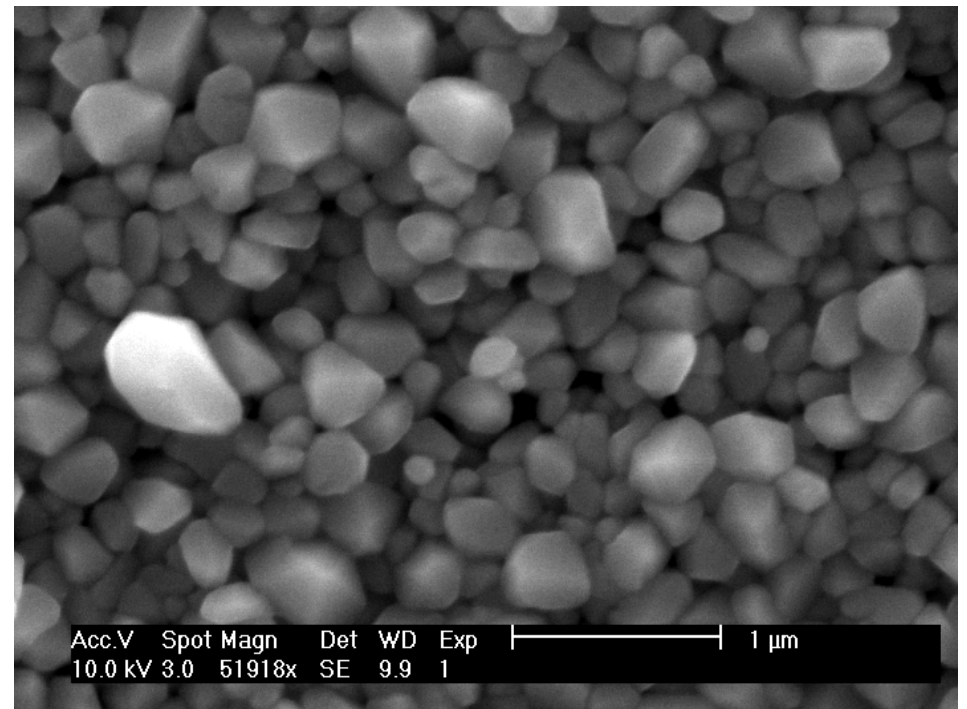
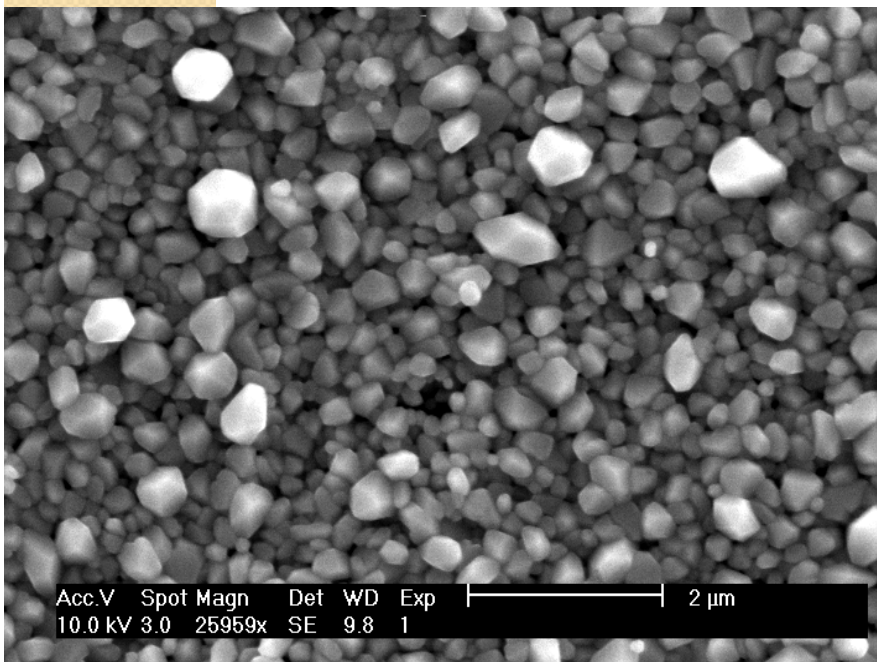
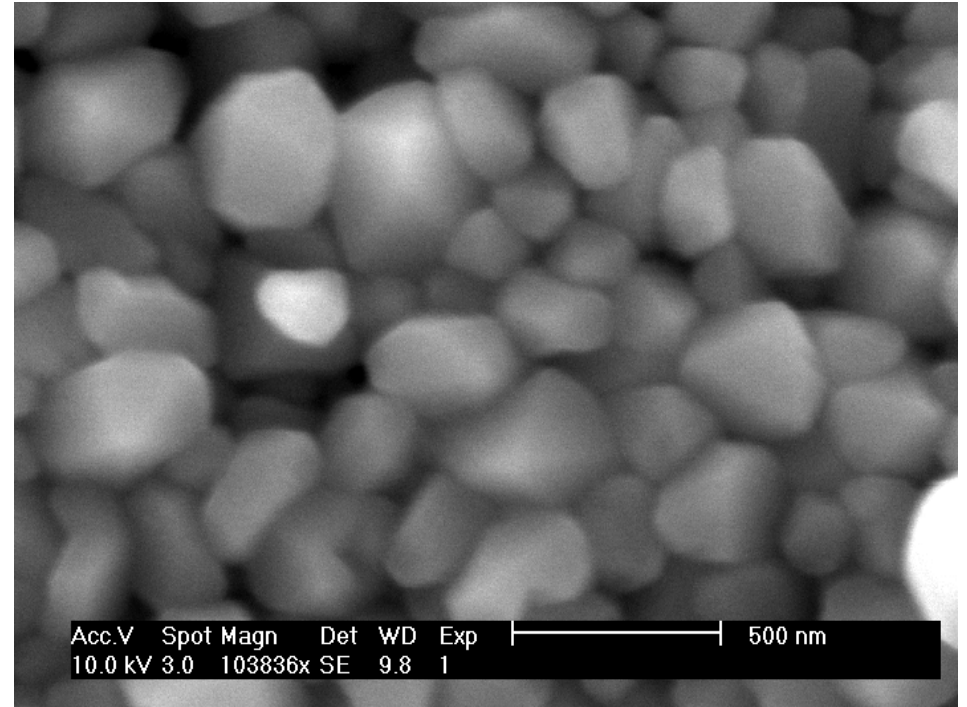
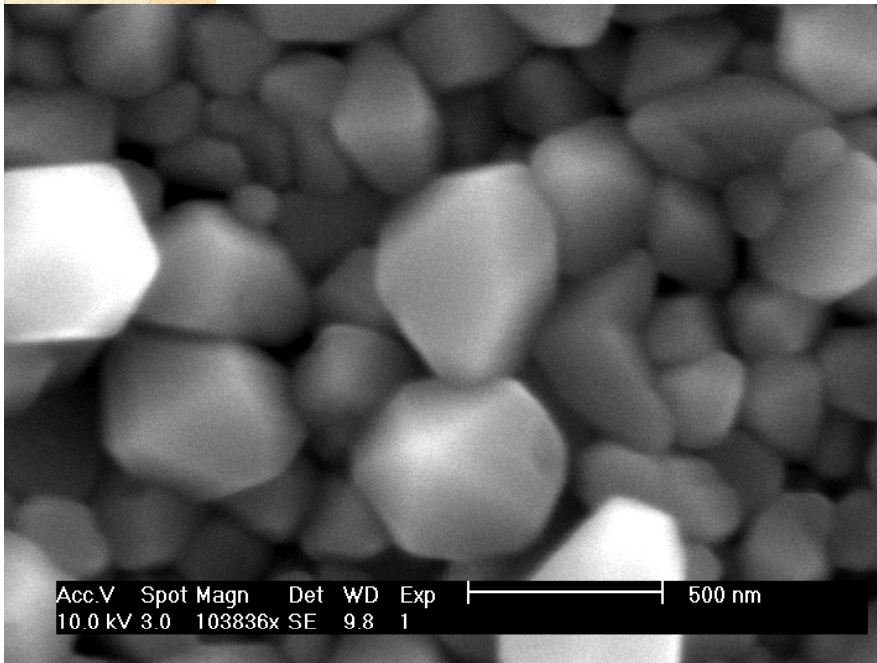


600 °C, 6h, 25mg S



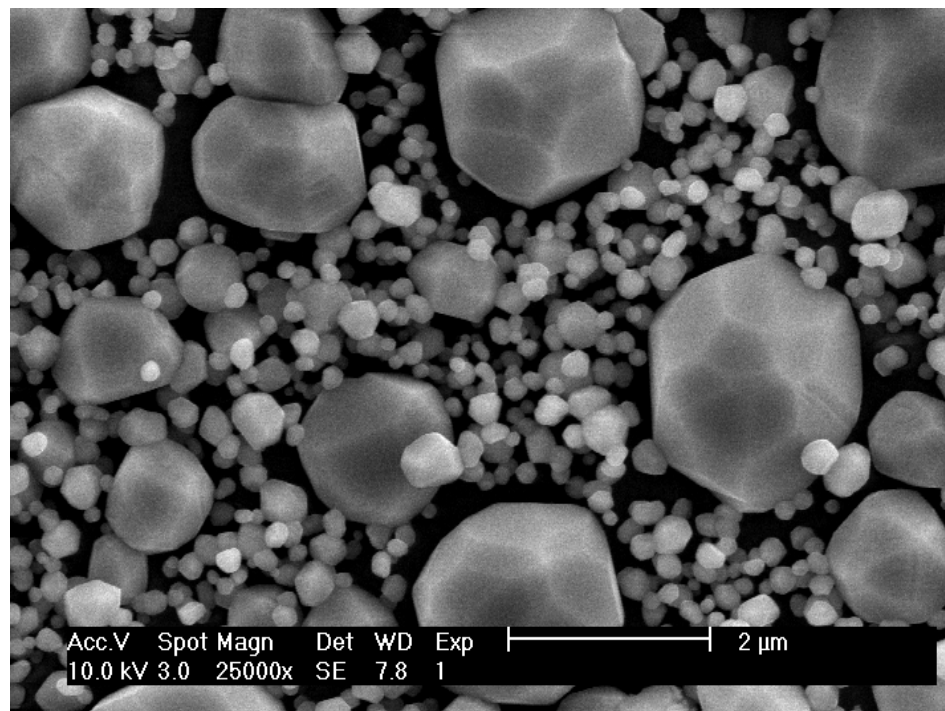
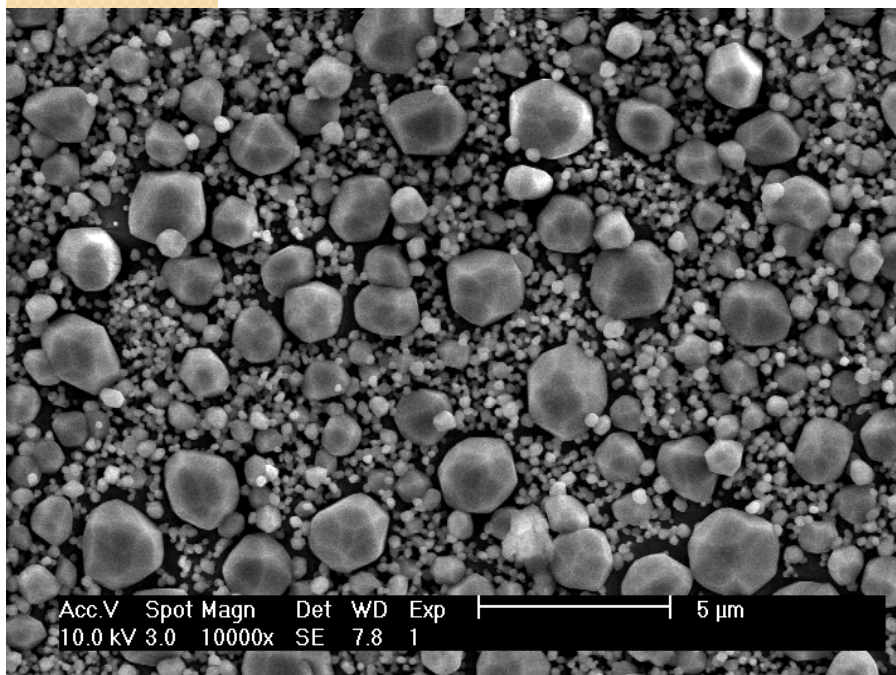
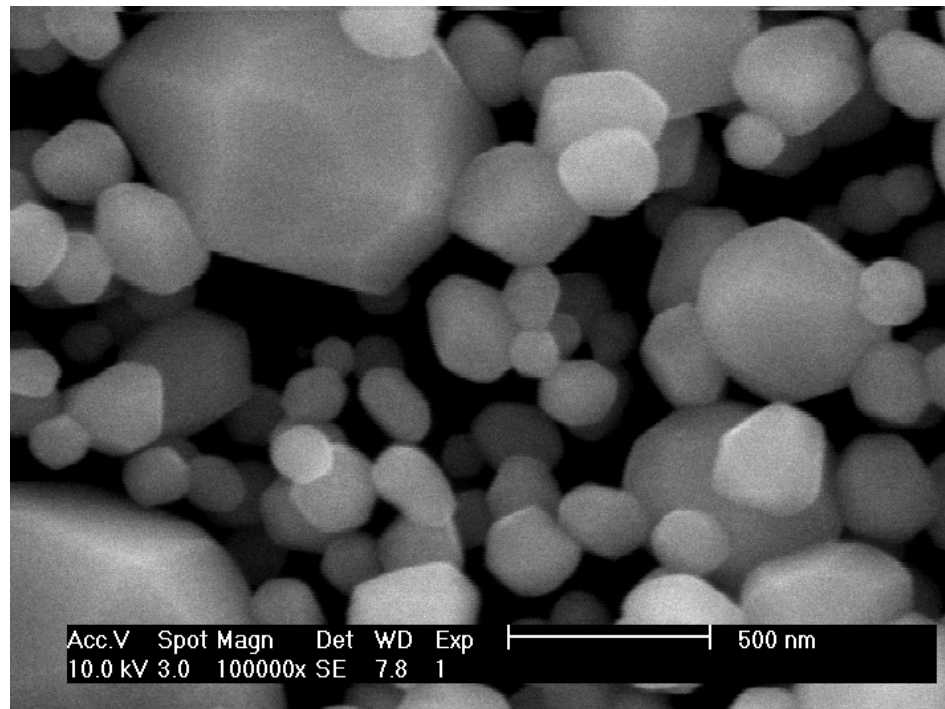
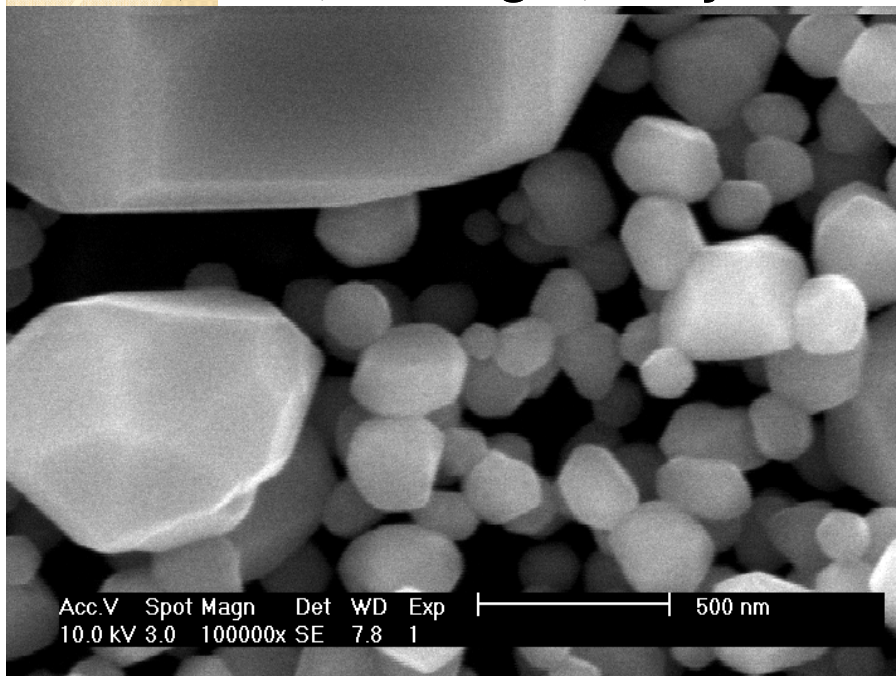


15mg S, 550 °C, 6hr, 7layers





550 °C, 6hrs, 100 mg S, 7 layers





# Future Work

- Determining chemical composition
  - Thermogravimetric Analysis
  - Differential Scanning Calorimetry
  - Mass Spectrometry
  - Powder X-Ray Diffraction
  - Nuclear Magnetic Resonance
  - XPS – look for contaminants (O, C, N, Na)
- S<sub>2</sub> annealing: vary S content, temperature, total cook time
- H<sub>2</sub>S annealing: time, temperature
- Exchanging ethanolamine with ethylene glycol (less toxic)
- Alloying with Zn